



UNIVERSIDAD DE CASTILLA - LA MANCHA

GUÍA DOCENTE

1. General information

Course: LABORATORY BASIC OPERATIONS

Type: BASIC

Degree: 409 - CHEMISTRY

Center: 1 - FACULTY OF SCIENCE AND CHEMICAL TECHNOLOGY

Year: 1

Main language: Spanish

Use of additional languages:

Web site:

Code: 57305

ECTS credits: 6

Academic year: 2022-23

Group(s): 20 23

Duration: C2

Second language:

English Friendly: Y

Bilingual: N

Lecturer: AURELIA ALAÑON MOLINA - Group(s): 20 23				
Building/Office	Department	Phone number	Email	Office hours
San Alberto Magno/planta baja	Q. ANALÍTICA Y TGIA. ALIMENTOS	6702	aurelia.alanon@uclm.es	
Lecturer: ALFONSO ARANDA RUBIO - Group(s): 20 23				
Building/Office	Department	Phone number	Email	Office hours
Marie Curie/2ª planta	QUÍMICA FÍSICA	3484	alfonso.aranda@uclm.es	Tuesday, Wednesday and Thursday. 12-14h
Lecturer: BEATRIZ CABAÑAS GALAN - Group(s): 20 23				
Building/Office	Department	Phone number	Email	Office hours
Edificio Marie Curie (primer piso)	QUÍMICA FÍSICA	6239	beatriz.cabanass@uclm.es	monday 12 to 14h, friday 10 to 12 h and tuesday 16 to 18h
Lecturer: AGUSTIN LARA SANCHEZ - Group(s): 20 23				
Building/Office	Department	Phone number	Email	Office hours
Edificio San Alberto Magno	QUÍMICA INORG., ORG., Y BIOQ.	3499	agustin.lara@uclm.es	Tuesday and Thursday from 17:00 to 19:00 h.
Lecturer: MARIA REYES LOPEZ ALAÑON - Group(s): 20 23				
Building/Office	Department	Phone number	Email	Office hours
Marie Curie (segunda planta))	QUÍMICA FÍSICA	3453	reyes.lopez@uclm.es	Tuesday and Wednesday: 10-12 h Thursday: 17-19 h
Lecturer: M ^a ISABEL LOPEZ SOLERA - Group(s): 20 23				
Building/Office	Department	Phone number	Email	Office hours
Edificio San Alberto Magno (primer piso)	QUÍMICA INORG., ORG., Y BIOQ.	6282	mabel.lopez@uclm.es	Monday, Wednesday and Friday: 12 - 14 h
Lecturer: JUANA RODRIGUEZ FLORES - Group(s): 20 23				
Building/Office	Department	Phone number	Email	Office hours
S. Alberto Magno	Q. ANALÍTICA Y TGIA. ALIMENTOS	3444	juana.rflores@uclm.es	TUESDAY AND WEDNESDAY FROM 16 TO 19
Lecturer: MARIA DEL PRADO SANCHEZ VERDU - Group(s): 20 23				
Building/Office	Department	Phone number	Email	Office hours
S. Alberto Magno, 2ª planta	QUÍMICA INORG., ORG., Y BIOQ.	3465	maria Prado.sanchez@uclm.es	tuesday and wednesday from 11 a.m. to 1 p.m.
Lecturer: ANA SANCHEZ-MIGALLON BERMEJO - Group(s): 20 23				
Building/Office	Department	Phone number	Email	Office hours
Edificio San Alberto Magno	QUÍMICA INORG., ORG., Y BIOQ.	3407	ana.smigallon@uclm.es	Monday from 12:00 to 14:00

2. Pre-Requisites

There are no prerequisites for this subject although it is recommended that the student has studied chemistry in high school. It is advisable that the student is taking the subject of Fundamentals of Chemistry

3. Justification in the curriculum, relation to other subjects and to the profession

Basic Laboratory Operations, is a fundamentally experimental subject. In this subject it is intended that the student acquires good working practices in a laboratory, manipulating with responsibility and security the chemical products; he will begin to know the material, the instrumentation and the basic operations of a laboratory through a series of practical activities. Likewise, they will acquire the necessary skills to be able to correctly develop the practical experiments that will be required in other subjects of the curriculum and in their future professional life.

4. Degree competences achieved in this course

Course competences

Code	Description
CB01	Prove that they have acquired and understood knowledge in a subject area that derives from general secondary education and is appropriate to a level based on advanced course books, and includes updated and cutting-edge aspects of their field of knowledge.
CB03	Be able to gather and process relevant information (usually within their subject area) to give opinions, including reflections on relevant social, scientific or ethical issues.
E01	Understand and use chemical terminology, nomenclature, conventions and units
E02	Deduce the variation of the properties of the chemical elements according to the Periodic Table
E03	Handle chemicals safely and with respect to the environment
E07	Relate macroscopic properties with those of atoms, molecules and non-molecular chemical compounds
E15	Know how to handle the standard chemical instrumentation and be able to elaborate and manage standardized procedures of work in the laboratory and chemical industry
G01	Know the principles and theories of Chemistry, as well as the methodologies and applications characteristic of analytical chemistry, physical chemistry, inorganic chemistry and organic chemistry, understanding the physical and mathematical bases that require
G02	Be able to gather and interpret data, information and relevant results, obtain conclusions and issue reasoned reports on scientific, technological or other problems that require the use of chemical tools
T03	Proper oral and written communication
T05	Organization and planning capacity

5. Objectives or Learning Outcomes

Course learning outcomes

Description

Encourage and promote in the student all those values and attitudes inherent to scientific activity.

Know the basic concepts and principles of Chemistry, so that the essential foundations are established so that they can successfully face the study of the different branches of the discipline.

Homogenize the knowledge of Chemistry already acquired by students in Secondary School courses and complete certain aspects that have not been previously studied with the necessary depth.

Achieve that the student acquires the basic terminology of Chemistry and knows how to use it, as well as being able to establish relationships between the different concepts.

Know and correctly handle the different units.

Learn to work autonomously in a laboratory and know how to interpret the experimental results obtained.

Additional outcomes

Handle and treat properly chemical reagents and their residues.

Learn the handling of laboratory material as well as different basic measuring instruments in a chemical laboratory.

Learn how to write a laboratory notebook and prepare a report on the activities carried out and the results obtained.

Skillfully perform basic laboratory operations

Knowing and complying with safety regulations in a laboratory.

6. Units / Contents

Unit 1: Security and waste management

Unit 2: Experimental introduction to basic chemical laboratory techniques

Unit 3: Laboratory material handling

Unit 4: Obtaining and analyzing results

Unit 5: Management of bibliographic search databases, calculation and presentation programs

ADDITIONAL COMMENTS, REMARKS

The contents of the course will be developed through a series of practical activities in which the different aspects and basic laboratory operations will be worked on.

7. Activities, Units/Modules and Methodology

Training Activity	Methodology	Related Competences	ECTS	Hours	As	Com	Description
Laboratory practice or sessions [ON-SITE]	Practical or hands-on activities	CB01 CB03 E01 E02 E03 E07 E15 G01 G02 T03 T05	2.56	64	Y	Y	Practical activities in which the different aspects and basic laboratory operations will be worked on. As it is an experimental activity of compulsory and not recoverable (according to Art. 4.3.f of the student's evaluation regulations) the realization of the practices will be compulsory in accordance with the official program of the subject.
Practicum and practical activities report writing or preparation [OFF-SITE]	Guided or supervised work	CB01 CB03 E01 G01 G02 T03	1.28	32	Y	Y	Creation of the workbook and related issues
Study and Exam Preparation [OFF-SITE]	Self-study	CB01 CB03 E01 G01 G02 T05	1.6	40	N	-	Autonomous work of the student to reach the competences
Final test [ON-SITE]	Assessment tests	CB01 CB03 E01 G01 G02 T03 T05	0.08	2	Y	Y	Written final evaluation test
Group tutoring sessions [ON-SITE]	Group tutoring sessions	E01 E03 E07 G01 G02	0.48	12	N	-	Resolution of doubts

Total:	6	150
Total credits of in-class work: 3.12	Total class time hours: 78	
Total credits of out of class work: 2.88	Total hours of out of class work: 72	

As: Assessable training activity

Com: Training activity of compulsory overcoming (It will be essential to overcome both continuous and non-continuous assessment).

8. Evaluation criteria and Grading System			
Evaluation System	Continuous assessment	Non-continuous evaluation*	Description
Laboratory sessions	60.00%	60.00%	Personal work in the laboratory will be evaluated: preparation of activities, order, cleanliness, compliance with safety standards, preparation of the laboratory notebook, carrying out calculations, carrying out practices, obtaining data and discussing results.
Final test	40.00%	40.00%	There will be a theoretical-practical exam on the activities carried out in the laboratory.
Total:	100.00%	100.00%	

According to art. 6 of the UCLM Student Evaluation Regulations, it must be provided to students who cannot regularly attend face-to-face training activities the passing of the subject, having the right (art. 13.2) to be globally graded, in 2 annual calls per subject, an ordinary and an extraordinary one (evaluating 100% of the competences).

Evaluation criteria for the final exam:

Continuous assessment:

The course will be passed when a minimum grade of 5.0 is obtained. It will be necessary to obtain a minimum grade of 4.0 in the final exam in order to average with the programmed activities.

Non-continuous evaluation:

The same criteria of the continuous evaluation are maintained, considering that the course includes practical activities of obligatory and not recoverable overcoming according to Art. 4.3.f of the student's evaluation regulations.

Specifications for the resit/retake exam:

The same criteria of the ordinary evaluation are maintained.

Specifications for the second resit / retake exam:

There will be a laboratory and content exam of the programmed practices, provided that the student has completed and passed the practical laboratory activities.

9. Assignments, course calendar and important dates	
Not related to the syllabus/contents	
Hours	hours
Practicum and practical activities report writing or preparation [AUTÓNOMA][Guided or supervised work]	32
Study and Exam Preparation [AUTÓNOMA][Self-study]	40
Final test [PRESENCIAL][Assessment tests]	2
General comments about the planning: Consult the weekly timetable of the course and the internship calendar published on the Virtual Campus.	
Unit 1 (de 5): Security and waste management	
Teaching period: 16-01-23 to 8-02-23	
Group 20:	
Initial date: 16-01-2023	End date: 19-01-2023
Group 23:	
Initial date: 20-01-2023	End date: 08-02-2023
Comment: PREPARATION OF SOLUTIONS Laboratory practicals [PRESENT] [Practical] 5 h STANDARDIZATION OF SOLUTIONS. ACID-BASE TITRATIONS. Laboratory practicals [PRESENT] [Practical] 5 h REACTIONS IN TEST TUBE Laboratory practices [PRESENT] [Practical] 5 h FILTRATION AND CRYSTALLIZATION TECHNIQUES. Laboratory practices [PRESENT] [Practical] 10 h QUALITATIVE ANALYSIS OF IONS. Laboratory practicals [PRESENT] [Practical] 5 h EXTRACTION TECHNIQUES. Laboratory practices [PRESENT] [Practical] 5 h SUBLIMATION Laboratory practicals [PRESENT] [Practical] 5 h SIMPLE AND FRACTIONAL DISTILLATION Laboratory practicals [PRESENT] [Practical] 5 h VACUUM DISTILLATION Laboratory practical [PRESENT] [Practical] 5 h DETERMINATION OF MELTING POINTS Laboratory practical [PRESENT] [Practical] 5 h THIN LAYER CHROMATOGRAPHY Laboratory practical [PRESENT] [Practical] 5 h CHEMICAL KINETICS Laboratory practicals [PRESENT] [Practical] 5 h ELECTROCHEMISTRY: VOLTAIC PILE Laboratory practice [PRESENT] [Practical] 5 h	
Global activity	
Activities	hours
Final test [PRESENCIAL][Assessment tests]	2
Practicum and practical activities report writing or preparation [AUTÓNOMA][Guided or supervised work]	32
Study and Exam Preparation [AUTÓNOMA][Self-study]	40
Total horas: 74	

10. Bibliography and Sources						
Author(s)	Title/Link	Publishing house	Citv	ISBN	Year	Description
J. Martínez Urrega y col.	Experimentación en química general	Thomson		84-9732-425-0	2006	
M.J. Rodríguez Yunta, F. G.	Curso experimental en Química Orgánica	Síntesis			2008	
Petrucci, Ralph H.	General chemistry: principles and modern applications	Prentice Hall Pearson-		0-13-014329-4	2002	

Petrucci, Ralph H.	Química general	Prentice Hall	978-84-205-3533-3	2010
Hill, Graham	Chemistry in context: laboratory manual	Nelson Thornes	0-17-448307-4	2001
V. Semishi	Prácticas de Química General Inorgánica	MIR		2009
Petrucci-Harwood-Hearing	Química General			
Szafran, Zvi	Microscale general chemistry laboratory: with selected macro	John Wiley & Sons	0-471-62114-5	1993